Low Temperature Joining Methods

•	Brazing	A low melting point metal is melted, drawn into the space between two solid surfaces by capillary action and allowed to solidify.
•	Soldering	A low melting point metal is melted, drawn into the space between two solid surfaces by capillary action and allowed to solidify.
•	Adhesive Joining	Uses a polymeric resin which fills the space between the surfaces to be joined.

Brazing

- Heat --- joining of metals through heat
- Filler material, whose temperature > 425 °C, but less than the melting point of the base metal

Difference in welding and brazing:

- 1. composition of brazing alloy different from base metal
- 2. strength of the brazing alloy different from base metal
- 3. melting point of brazing alloy different from base metal
- 4. bonding requires capillary action

Advantages:

- 1. ideal for dissimilar metals (all metals can be joined)
- 2. quick and economical (less heat)
- 3. thinner and more complex assemblies can be joined successfully (temperature is less, little HAZ, less warping, distortion)
- 4. mass production of delicate assemblies (automation)

Disadvantages:

- 1. reheating of brazing material can cause the destruction of joint
- 2. low joint strength

Brazing joints: butt, scarf, lap, or shear

- strong metallurgical bond at the interfaces
- clean surfaces; good wetting (function of surface tension); good fluidity (function of filler metal, temperature)
- bind strength (function of material, processing, design, clearance)
- apply some pressure

Brazing metals:

- copper and its alloys
- silver and its alloys
- aluminum alloys

<u>Fluxes:</u> (e.g. borax)

role is:

- dissolving oxide that may be on the surface prior to heating
- prevent the formation of oxides during heating
- lowering the surface tension

Methods of brazing:

- 1. torch brazing oxyacetylene or oxyhrogen or other gas flame (good for repair work)
- 2. furnace brazing controlled atmosphere or vacuum, preload the brazing metal
- 3. salt bath brazing parts are heated by dipping in bath of molten salt
- 4. dip brazing assemblies are immersed in bath of molten brazing metal
- 5. induction brazing high frequency induction current
- 6. resistance brazing under pressure between two electrodes

Note: before application, the surface must be cleaned. Some flux are corrosive and must be removed after brazing.

Soldering

Soldering is a brazing-type operation where the filler metal has a melting temperature below 425 °C. bond strength is relatively low, the joining being the result of adhesion between the solder and the parent metal.

Solder metal:

- alloys of lead and tin
- tin-aluminum alloys are used in electric applications
- tin-zinc, cadnium-zinc, or aluminum-zinc is often used to solder aluminum
- lead-silver or cadnium silver --- high temperature application
- indium-tin alloys are used when joining metal to glass

Soldering fluxes:

- soldering fluxes are not intended to, and will not, remove any appreciable amount of contamination
- two types: corrosive (muriatic acid; mixture of Zn and ammonium chlorides, nickel, aluminum, copper, brass, zinc, steel) and non-corrosive (rosin in alcohol, aniline phosphate)

Heating for soldering:

Any method that is used for brazing can be used for soldering.

- furnace and salt-bath heating seldom used
- dip soldering for soldering wire ends, electronic work, automobile radiators and tinning
- induction heating where identical parts are soldered

Design and strength of soldered joint:

- seldom develop shear strength in excess of 1.71 MPa
- never use butt joints
- hold together until solidification, otherwise full of cracks

Flux removal:

- to protect from corrosion
- for appearance

Adhesive Bonding

Potential areas:

- automobile industry
- aircraft industry

Adhesives:

- thermoplastics
- thermosetting resins
- artificial elastomers
- ceramics

Structural adhesives:

are composite systems with several components - liquids, pastes, solids, pellets, cartridges, tapes, or films.

Common adhesives:

- 1. epoxies
- 2. cyanoacrylates
- 3. anaerobics
- 4. acrylics
- 5. urethane or polyurethane
- 6. silicones
- 7. high-temperature adhesives
- 8. holt melts

Joint design and preparation:

two types:

- 1. continuous surface
- 2. core-to-face

Quality adhesive-bonded joint:

4 steps to obtain a quality joint:

- 1. cleaning
- 2. etching (to provide maximum wetting characteristics)
- 3. rinsing
- 4. drying

Advantages:

- 1. any material or combination or materials can be joined
- 2. very thin and quite delicate materials (e.g. foils) can be joined to each other or heavier sections
- 3. good load distribution and fatigue resistance (continuous bond)
- 4. smooth contours are also obtainable (no holes, rivets, bolts)
- 5. thermal and electrical insulation
- 6. cost savings (simplified machining, assembly, reduced finishing, elimination of mechanical fasteners, absence of high skilled labor)

Disadvantages:

- 1. most adhesive are not stable above 177 $^{\circ}\mathrm{C}$
- 2. it is difficult to determine quality of joint (with NDT methods)
- 3. surface preparation, adhesive preparation and curing are critical
- 4. life expectancy is hard to predict
- 5. assembly time greater (sometimes)
- 6. some contain objectionable chemicals